

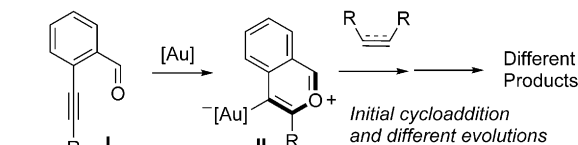
Gold Catalysis

International Edition: DOI: 10.1002/anie.201602948
German Edition: DOI: 10.1002/ange.201602948Gold(I)-Catalyzed Generation of the Two Components of a Formal [4+2] Cycloaddition Reaction for the Synthesis of Tetracyclic Pyrano[2,3,4-*de*]chromenes

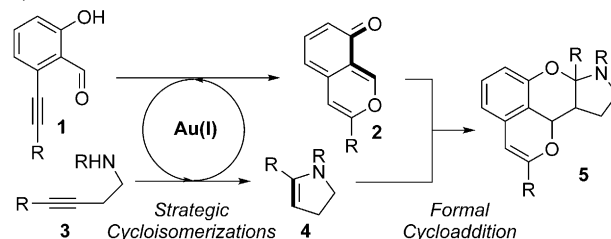
Tamara Arto, Francisco J. Fañanás,* and Félix Rodríguez*

Abstract: *Ortho*-alkynylbenzaldehydes have been widely used to generate isochromenylium derivatives through gold-catalyzed cycloisomerization. These isochromenylium derivatives have been exploited as formal diene derivatives for reactions with different dienophiles. Herein, we describe the behavior of *ortho*-alkynylsalicylaldehydes, a particular case of *ortho*-alkynylbenzaldehydes. The gold-catalyzed cycloisomerization of *ortho*-alkynylsalicylaldehydes delivers an unusual heterodiene derivative that reacts with electron-rich alkenes through a formal [4+2] cycloaddition. In this reaction, both the diene and dienophile are generated in situ through gold-catalyzed cycloisomerization of appropriate alkynamines or alkynols. This reaction was used to synthesize complex tetracyclic pyrano[2,3,4-*de*]chromenes from two very simple starting materials (an *ortho*-alkynylsalicylaldehyde and an alkynamine or alkynol) with complete atom economy and with selective formation of bonds, cycles, and stereocenters.

The tremendous advances in the context of gold catalysis witnessed during the last decade have brought about a revolution in organic synthesis.^[1] The cycloisomerization of alkyne-containing molecules is one very useful process where gold catalysts have become unbeatable. For example, the cycloisomerization of *ortho*-alkynylbenzaldehydes **1** is a powerful strategy to obtain the versatile synthetic intermediates isochromenyliums **II** (Scheme 1a).^[2] These intermediates are usually employed as the heterodiene partner in processes initiated by [4+2] cycloaddition reactions with appropriate dienophiles.^[3] In this context, and following our interest in the development of new reactions based on catalytic cycloisomerization reactions,^[4] we envisioned that the gold-catalyzed cycloisomerization of alkynylsalicylaldehydes **1**, a particular type of *ortho*-alkynylbenzaldehydes **I**, would deliver the isochromanone derivatives **2** (Scheme 1b).^[5] Interestingly, these compounds contain in their structures a 1,4-heterodiene moiety different from that obtained from simple *ortho*-alkynylbenzaldehydes **I** (isochromenyliums **II** are obtained from **I** but isochromanones **2** from **1**). To our knowledge, the presence of the mentioned heterodiene moiety in **2** has not been synthetically exploited. In this context, we thought that intermediates **2** could react

a) Conventional reactivity of *o*-alkynylbenzaldehydes (Ref. [3])

b) This work:



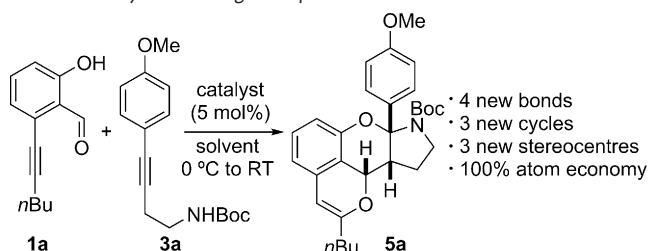
Scheme 1. The conventional reactivity of *ortho*-alkynylbenzaldehydes **1** and our proposed reaction.

with electron-rich alkene derivatives through a formal [4+2] cycloaddition. Our goal was not only to develop this new formal [4+2] cycloaddition reaction but also to generate both precursors, the heterodiene and the dienophile, in situ from simple reagents. We envisaged that a unique gold catalyst could promote the synchronized formation of both reagents, the dienes **2** from alkynylsalicylaldehydes **1** and the dienophiles **4** from alkynamines **3**, through cycloisomerization reactions.^[6] Subsequent [4+2] cycloaddition between **2** and **4** would then deliver interesting polycyclic compounds **5** in an apparently straightforward way from very simple starting materials. It should be noted that although the in situ generation of dienes or dienophiles for subsequent cycloaddition has been reported,^[7] to our knowledge, this concept of producing both reagents at the same time through the action of a single catalyst has not been fully exploited.

Our initial investigation was performed with *ortho*-alkynylsalicylaldehyde **1a** and the *N*-Boc-protected alkynamine **3a** as model substrates (Table 1). The first experiment was carried out in dichloromethane as the solvent and in the presence of the cationic gold(I) complex generated in situ from 5 mol % of [AuCl(PPh₃)] and 5 mol % of AgOTf (Table 1, entry 1). Pleasingly, this reaction afforded the desired polycyclic compound **5a** in 40 % yield, thus demonstrating the viability of our proposal. However, under these conditions, an equimolecular mixture of two diastereoisomers was obtained. A brief screening of solvents showed that the diastereoselectivity of the reaction could be easily improved by using toluene or tetrahydrofuran (THF) as solvents

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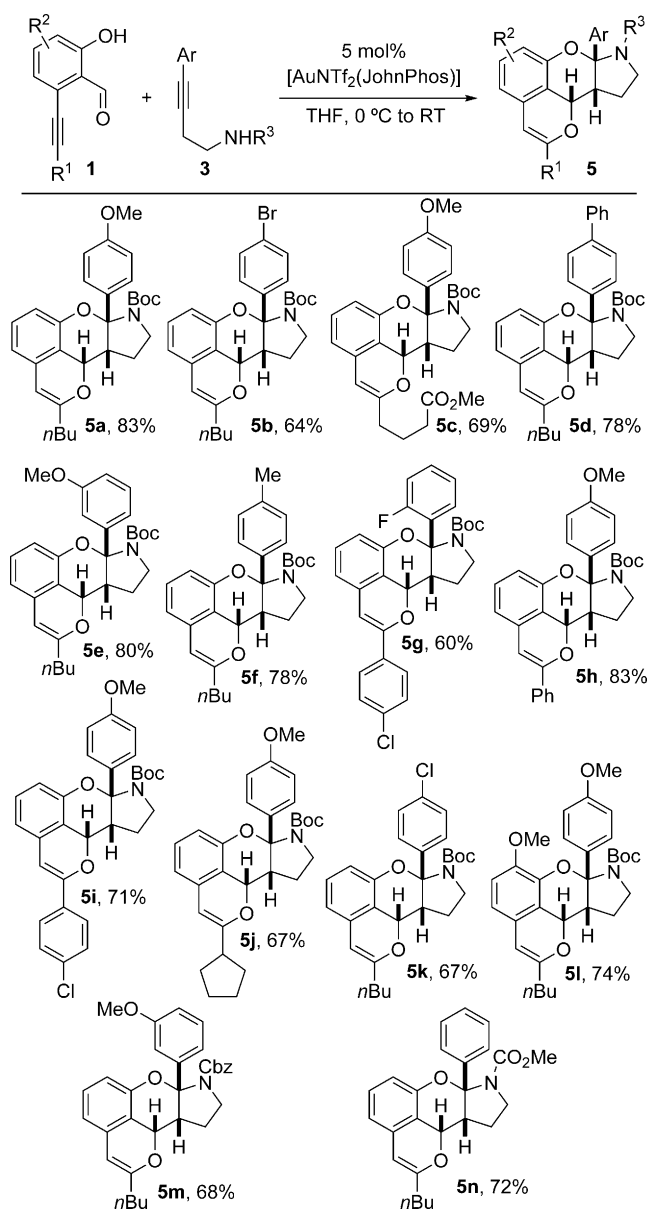
Table 1: Catalyst screening and optimization of the reaction conditions.^[a]


Entry	Catalyst	Solvent	Yield (d.r.) ^[a]
1	[AuCl(Ph ₃ P)]/ AgOTf	CH ₂ Cl ₂	40% (1:1)
2	[AuCl(Ph ₃ P)]/ AgOTf	toluene	52% (> 20:1)
3	[AuCl(Ph ₃ P)]/ AgOTf	THF	63% (> 20:1)
4	[AuNTf ₂ (JohnPhos)]	THF	83% (> 20:1)
5	AgOTf	THF	—
6	PtCl ₄	THF	—

[a] The reactions were performed by adding the catalyst (5 mol%) to a solution of **1a** (1 equiv) and **3a** (1.1 equiv) at 0 °C and then allowing stirring the reaction at room temperature for 6 hours. [b] Yield of isolated product based on **1a**. The diastereoisomeric ratio was determined by ¹H NMR analysis of the crude reaction mixture.

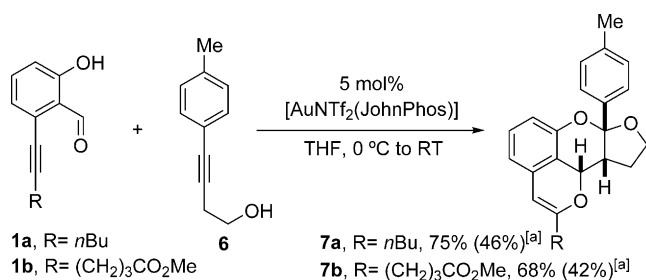
(Table 1, entries 2 and 3). Finally, we observed that more reliable results in terms of yield and selectivity were obtained by using the preformed cationic gold(I) catalyst [AuNTf₂-(JohnPhos)] in THF as a solvent (Table 1, entry 4). Other typical silver- and platinum-derived catalysts used in related cycloisomerization reactions of *ortho*-alkynylbenzaldehydes did not afford positive results (Table 1, entries 5 and 6). The intricate transformation observed in this reaction should be remarked upon at this point. A complex tetracyclic compound is obtained from two simple starting materials in a process in which four new bonds, three cycles, and three stereogenic centers (one of them on a quaternary carbon atom) are formed in a selective way and without the generation of any byproducts (100% atom economy). It should also be noted that the reactivity observed with *ortho*-alkynylsalicylaldehyde **1a** is different from the conventional reactivity of *ortho*-alkynylbenzaldehydes (such as **1** in Scheme 1) and electron-rich alkenes under gold catalysis.^[8]

With optimal reaction conditions in hand, we next examined the scope of this new gold-catalyzed polycyclization process. As illustrated in Scheme 2, this transformation proceeds at room temperature and in good to high yield with a wide range of *ortho*-alkynylsalicylaldehydes **1** and alkyamine derivatives **3**. Specifically, the reaction seems general with respect to the substitution of aldehydes **1**, and aromatic and aliphatic substituents at the R¹ position were accepted. *ortho*-Alkynylsalicylaldehydes **1**, which are substituted at the aromatic ring (R²), are also appropriate substrates for this reaction (**5l**). Regarding the alkyamines **3**, the reaction also seems general whenever the alkyne is substituted with an aryl group (alkyl substitution or unsubstituted alkyamines did not provide the desired products). Different protecting groups, Boc (**5a–l**), Cbz (**5m**) and CO₂Me (**5n**), on the nitrogen atom of the alkyamine derivatives **3** are accepted. Moreover, compounds **5** were formed as single diastereoisomers in all cases.

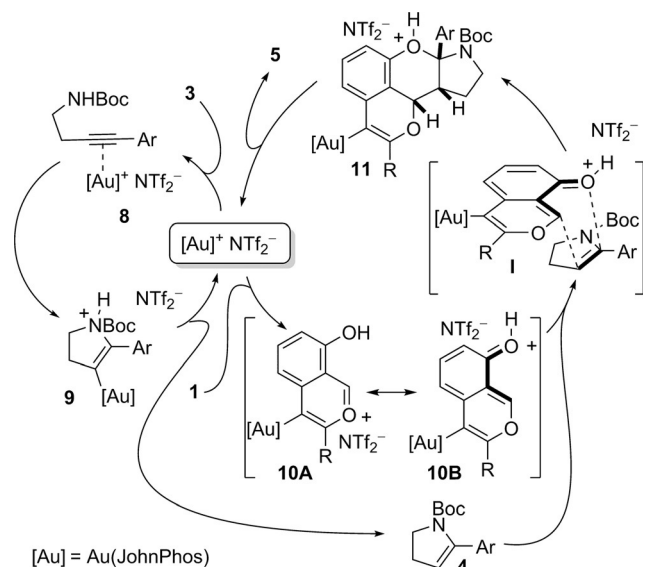
**Scheme 2.** Synthesis of chromeno[2,3-*b*]pyrrole derivatives **5**.

This new gold-catalyzed reaction is not limited to the use of alkyamines **3** as precursors of electron-rich alkene derivatives for coupling with the proposed heterodienes **2**. In this context, we also used the alkynol derivative **6** to generate a cyclic enol ether as the formal dienophile for the proposed cycloaddition reaction. The reaction of *ortho*-alkynylsalicylaldehydes **1a,b** with alkynol derivative **6** in the presence of 5 mol% of catalyst [AuNTf₂(JohnPhos)] in THF from 0 °C to room temperature afforded the furo[2,3-*b*]pyrano[2,3,4-*de*]chromene derivatives **7a,b** in good yield and as single diastereoisomers (Scheme 3). Although these furan derivatives **7**, once isolated, were stable compounds, we observed some degradation under the column chromatography purification conditions.

The mechanism proposed to explain the formation of products **5** is shown in Scheme 4. The reaction is initiated by



Scheme 3. Synthesis of furo[2,3-*b*]pyrro[2,3,4-*de*]chromene derivatives **7**. [a] Yield determined by ¹H-NMR using an internal standard, with the yield of isolated product after column chromatography given in parentheses.



Scheme 4. Proposed mechanism for the formation of compounds **5**.

coordination of the cationic gold complex to the triple bond of the starting alkynamine derivative **3** to form intermediate **8**. Intramolecular addition of the nitrogen atom to the external carbon of the triple bond generates **9**. Protodemetalation of the latter affords the enamine derivative **4** and releases the catalytic gold species.

The gold catalyst is also able to activate the other starting material, the *ortho*-alkynylsalicylaldehyde **1**, through coordination to the alkyne. This coordination promotes intramolecular addition of the oxygen of the aldehyde to form species **10** (Scheme 4). This intermediate may be represented as the isochromenylium **10A** or the isochromanone resonance form **10B**. The subsequent formal inverse-electron-demand [4+2] cycloaddition reaction between the activated electron-poor heterodiene shown in the isochromanone resonance form **10B** and the previously formed electron-rich enamine derivative **4** delivers the new intermediate **11**. A final protodemetalation step closes the second catalytic cycle to afford the final chromeno[2,3-*b*]pyrrole derivative **5** and regenerate the gold catalyst. The relative configuration observed at the new stereogenic centers in the final products **5** could be explained by the dienophile approaching the diene by following an *endo* trajectory as shown in **I** (*endo* refers to the orientation of the

nitrogen in dienophile **4** with respect to the diene **10B**).^[9] As shown, the global process could be considered a new example of self-relay one-pot catalysis because one catalyst promotes at least two mechanistically distinct reactions in a single flask.^[10]

In summary, we have developed a new gold-catalyzed cascade reaction of *ortho*-alkynylsalicylaldehydes and alkynamines (or alkynols) towards functionalized tetracyclic compounds in a process in which several bonds, cycles, and stereocenters are formed in a selective way. The process is operationally simple, 100% atom-economic, and further demonstrates the ability of gold catalysts to promote complex transformations that would be otherwise difficult to accomplish. The reaction proceeds through a double cycloisomerization reaction followed by a formal [4+2] cycloaddition. Interestingly, in this reaction, the gold catalyst promotes the *in situ* formation of both the diene and dienophile.

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